

# 3rd International Conference on Environmental Design, ICED2022

## Polymer-coated Nanoparticles used as agents for Enhanced Oil Recovery

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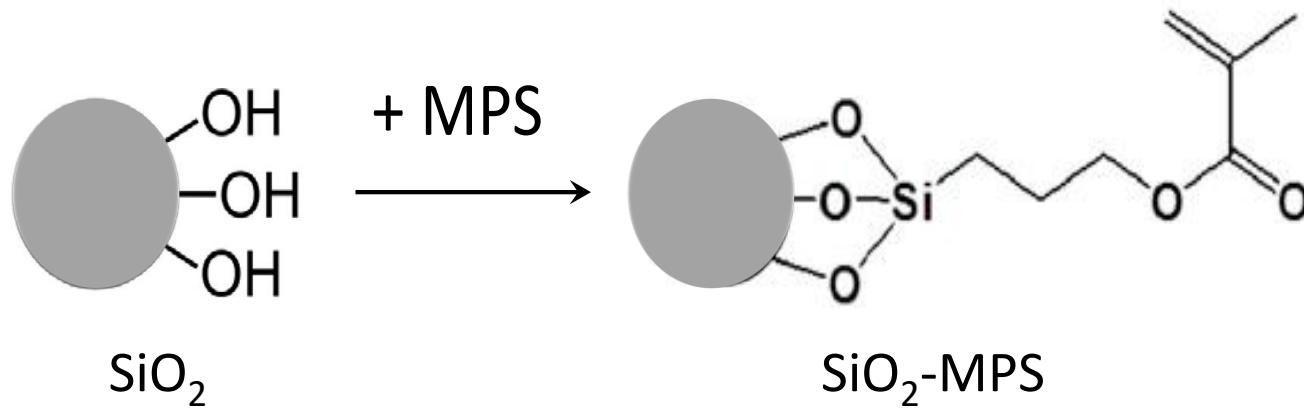
## Contents

- Polymer-coated nanoparticles (PNPs)
- PNPs Characterization
- Surface/Interfacial Tension and Rheology Characterization
- Displacement Tests

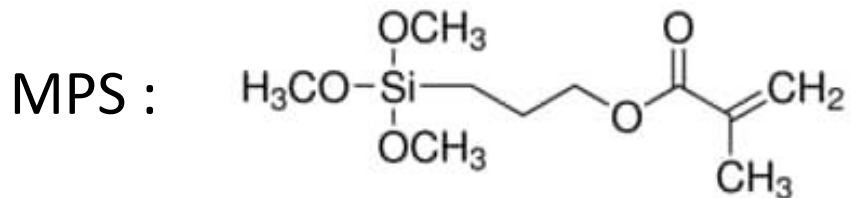
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## Polymer-coated nanoparticles (PNPs)

1) Functionalization of  $\text{SiO}_2$  NPs with 3-(trimethoxysilyl)-propyl methacrylate ( $\text{SiO}_2$ -MPS)



- ✓  $\text{SiO}_2$  NPs in toluene-sonication
- ✓ add MPS under vigorous stirring
- ✓ stirring and reflux at 100 °C for 24 h
- ✓ separated by centrifugation at 9000 rpm
- ✓ three “wash by EtOH/centrifuge” cycles
- ✓ dry in vacuum oven at 60 °C overnight

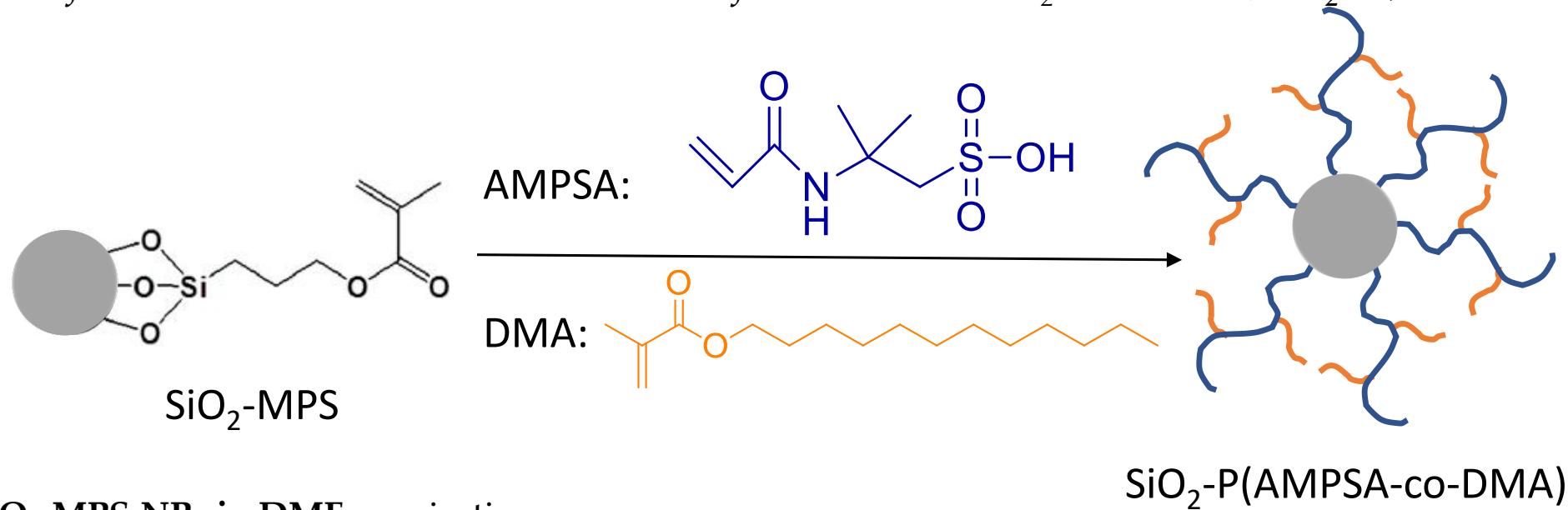


*3-(methacryloxy)propyltrimethoxysilane*

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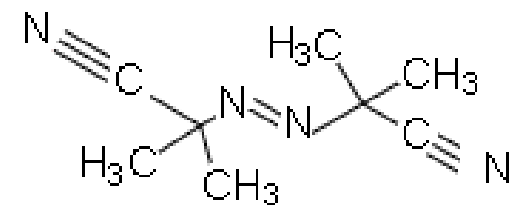
## Polymer-coated nanoparticles (PNPs)

2) Polymerization of AMPSA and DMA monomers onto the functionalized  $\text{SiO}_2$ -MPS NPs ( $\text{SiO}_2$ -P(AMPSA-co-DMA))



- ✓ dispersed  $\text{SiO}_2$ -MPS NPs in DMF - sonication
- ✓ add AMPSA and DMA monomers and initiator AIBN
- ✓ stirring under  $\text{N}_2$ , at 80 °C for 24 h
- ✓ separated by centrifugation at 11000 rpm
- ✓ three “wash by  $\text{H}_2\text{O}$ /centrifuge” cycles, dry in vacuum oven at 50 °C overnight

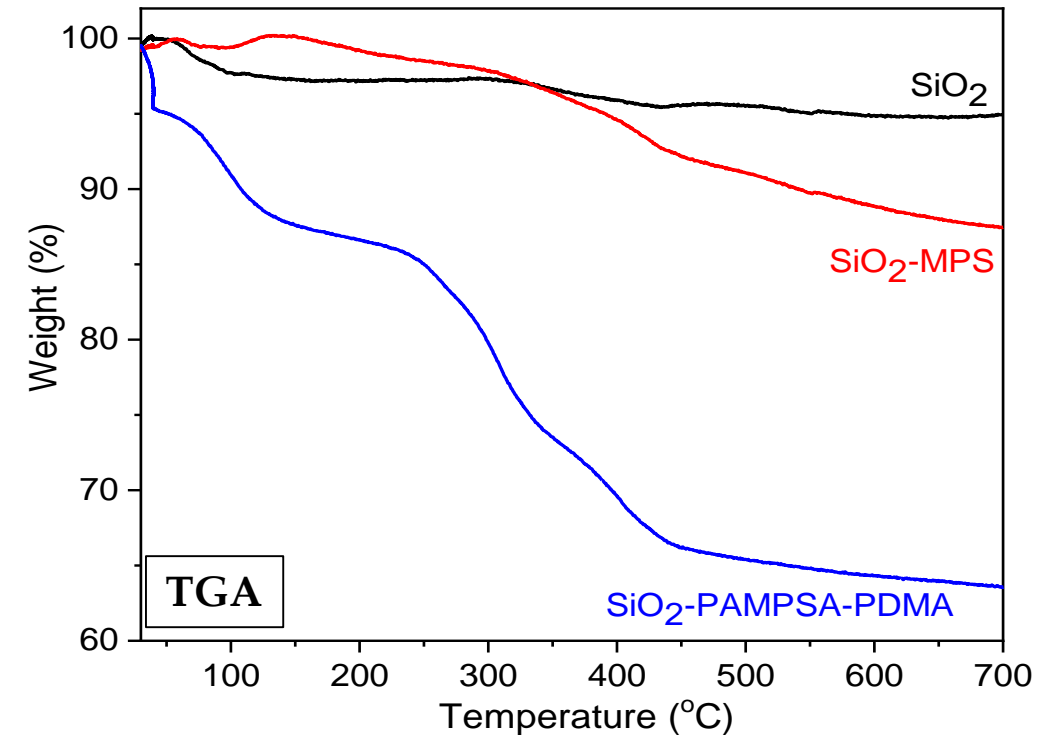
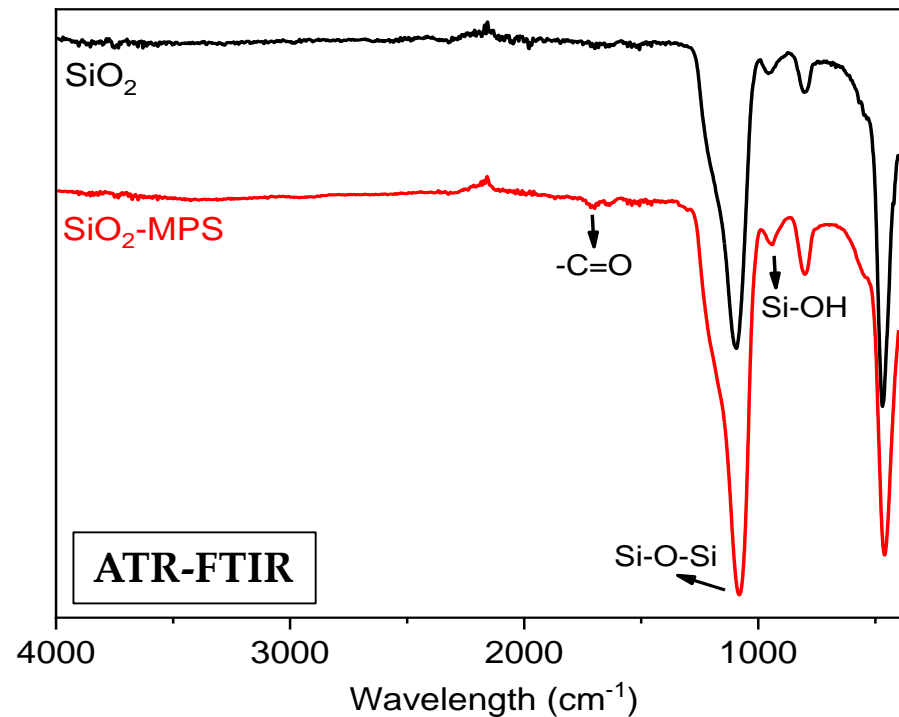
AIBN:



Azobisisobutyronitrile

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## PNPs Characterization



In both  $\text{SiO}_2$  and  $\text{SiO}_2\text{-MPS}$ :

1095 and 465  $\text{cm}^{-1}$ : asymmetric stretching vibration of **Si-O-Si** groups of silica

955  $\text{cm}^{-1}$ : vibration of **Si-OH** groups of silica

In  $\text{SiO}_2\text{-MPS}$  NPs:

1640  $\text{cm}^{-1}$ : new stretching vibration peaks of **C=C** groups from MPS

1715  $\text{cm}^{-1}$ : stretching vibration peaks of **C=O** groups from MPS

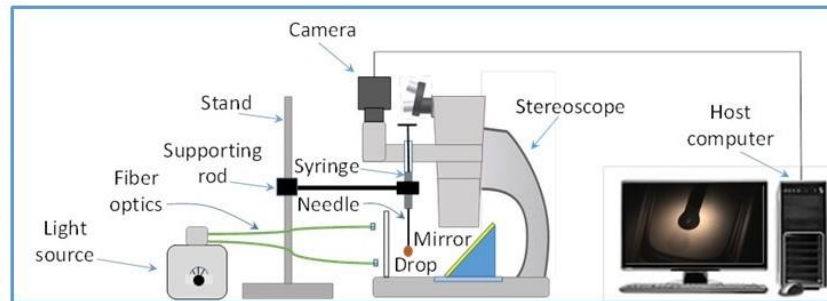
✓ Successful surface modification of the  $\text{SiO}_2$  NPs by MPS and at a next step by P(AMPSA-co-DMA)

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## Surface Tension Characterization

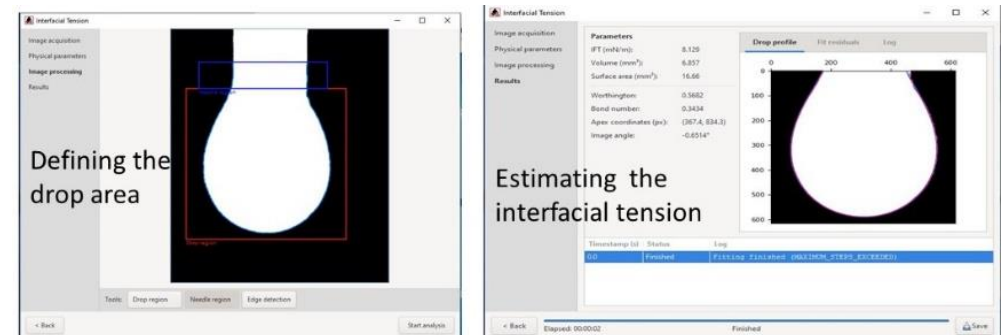
### Pendant Drop Method:

Dynamic Surface and Interfacial Tension of nano-colloid suspensions



analyzing the recorded interfacial configurations with the open access OpenDrop software of inverse modeling of Young-Laplace equation

*Experimental setup of pendant drop method*



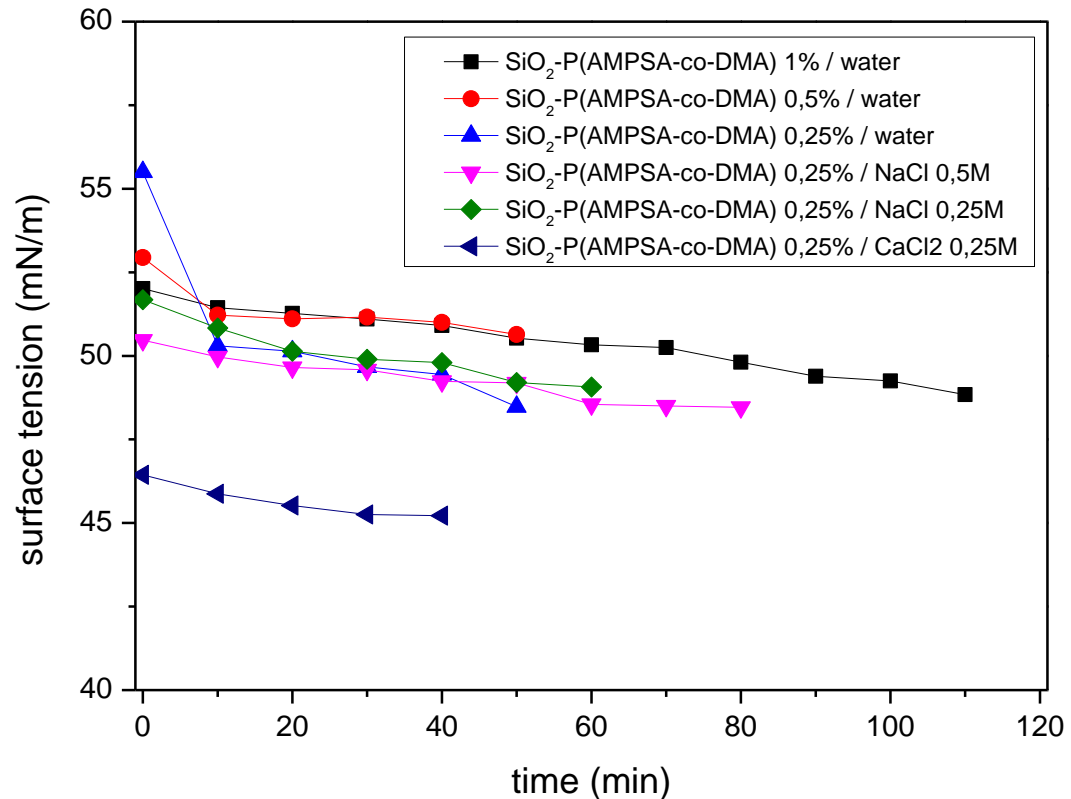
*Estimation of surface/interfacial tension by OpenDrop software*



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## Surface Tension Characterization

*Dynamic surface tension (ST) as function of time for various concentrations of SiO<sub>2</sub>-P(AMPSA-co-DMA) NPs in water and salt solutions (NaCl, CaCl<sub>2</sub>).*

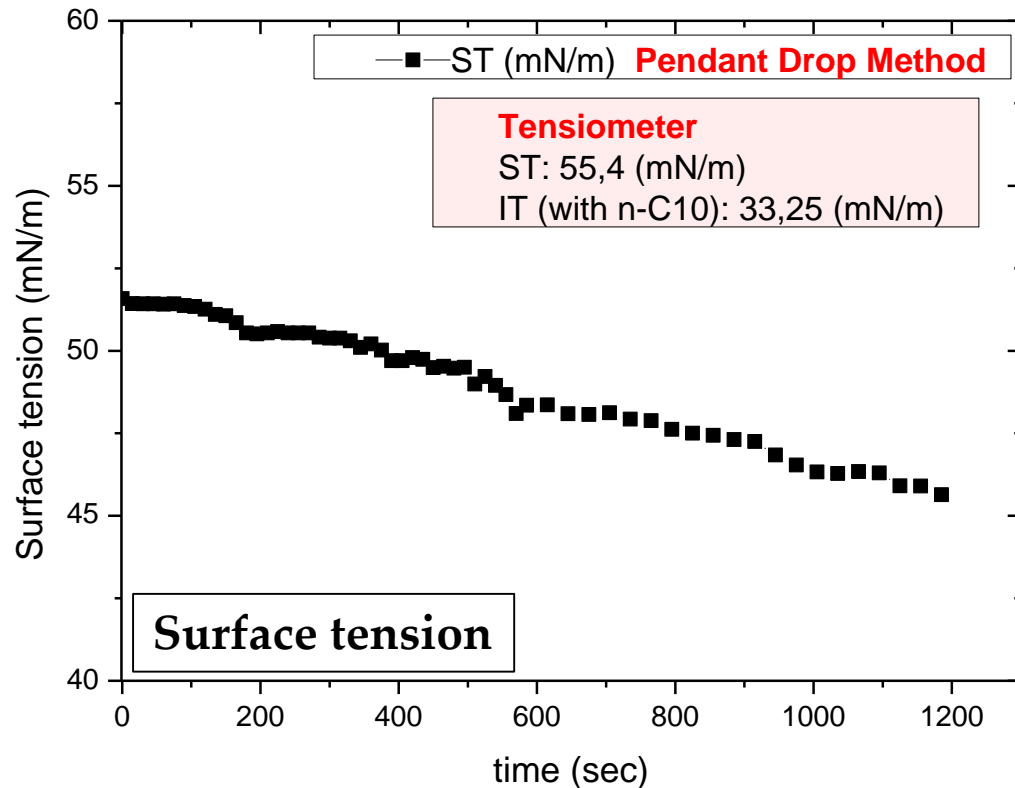


- Maximized reduction rate of ST of SiO<sub>2</sub>-P(AMPSA-co-DMA) NPs at PNP concentration equal to 0.25%
- with the addition of NaCl, the ST changes weakly
- with the addition of CaCl<sub>2</sub>, the ST drops significantly, due to the stronger electrostatic interactions of the divalent CaCl<sub>2</sub> with the P(AMPSA-co-DMA) polyelectrolyte and increased ionic strength

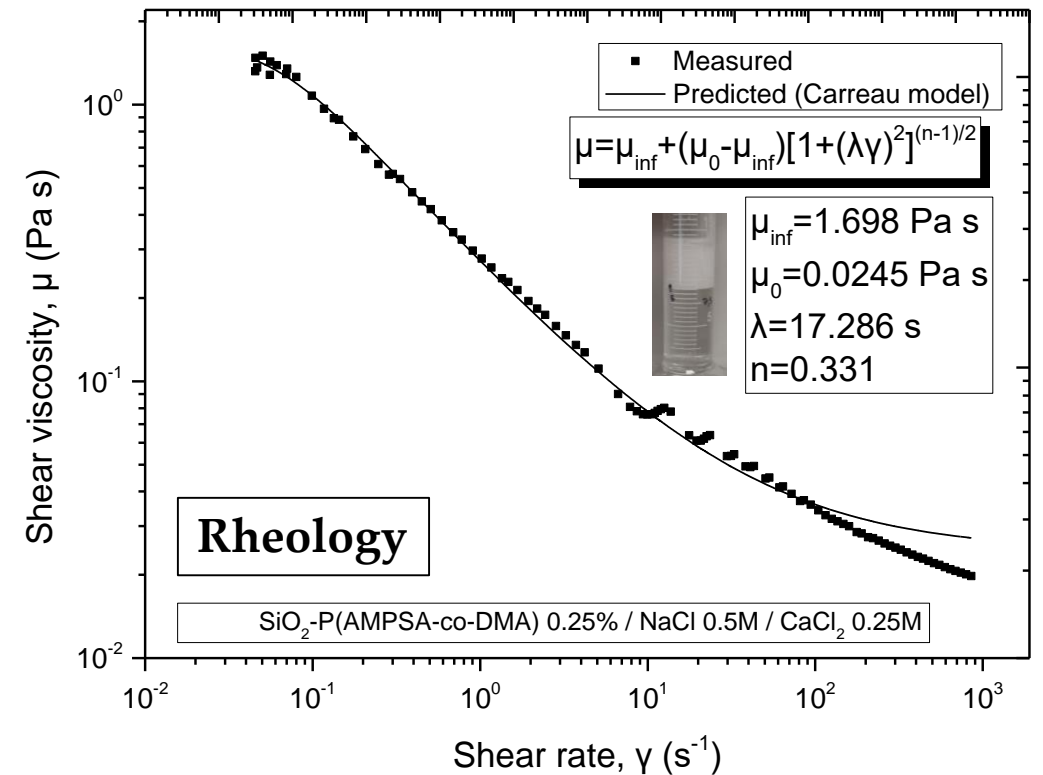
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## Surface Tension and Rheology Characterization

*PNP dispersion: SiO<sub>2</sub>-P(AMPSA-co-DMA 0.25% / NaCl 0.5M / CaCl<sub>2</sub> 0.25M*



respectable reduction rate of ST with time



The shear thinning rheology of the Pickering emulsion was fitted satisfactorily with the Carreau model

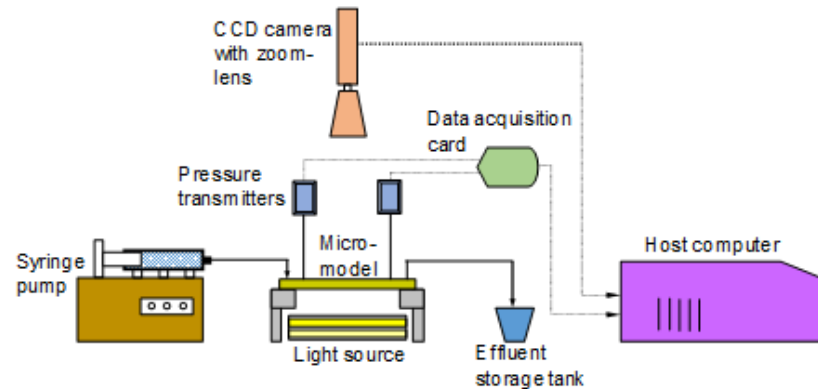


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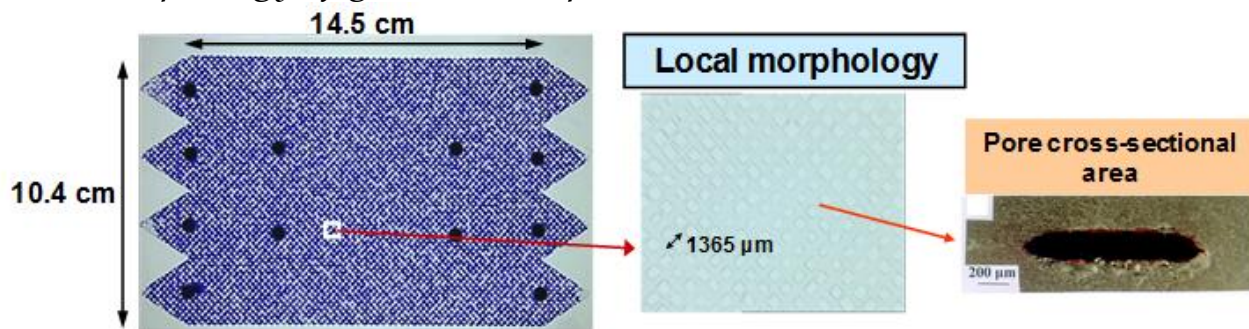
## Displacement Test

### Visualization EOR tests in a glass-etched pore network

(a) Schematic diagram of experimental setup.



(b) Morphology of glass-etched pore network model.



### Displacement Tests conducted on the glass micromodel

Transient responses of paraffin oil saturation for displacement tests,

- **brine** : aqueous solution of NaCl 0.5M/CaCl<sub>2</sub> 0.25M
- **displacing fluid in secondary imbibition:**

➤ **the PNP dispersion**

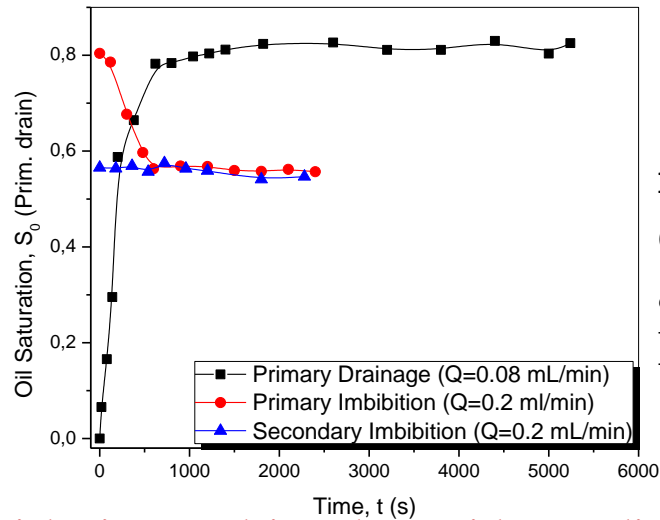
[SiO<sub>2</sub>-P(AMPSA-co-DMA) 0.25%/NaCl 0.5M/CaCl<sub>2</sub> 0.25M]

or

➤ **Pickering emulsion along with PNP dispersion**

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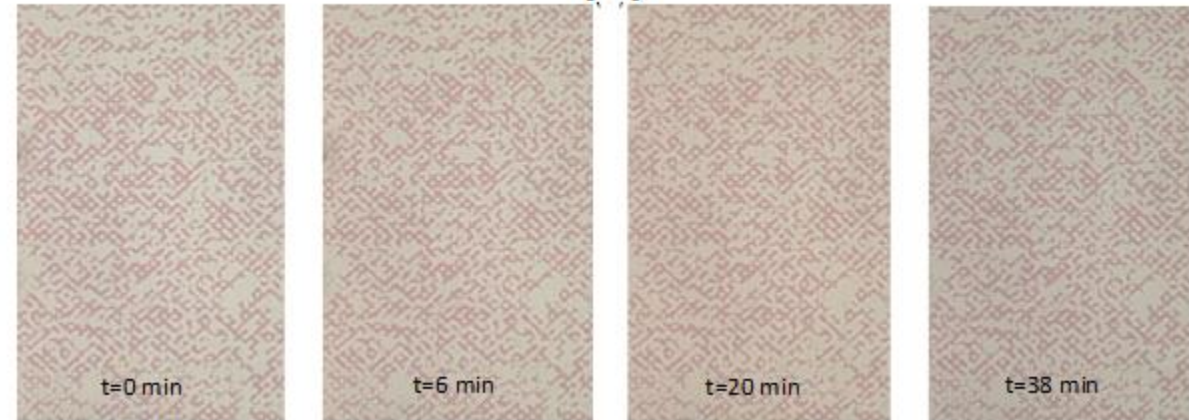
**(a) [SiO<sub>2</sub>-P(AMPSA-co-DMA) 0.25% / NaCl 0.5M / CaCl<sub>2</sub> 0.25M] dispersion**



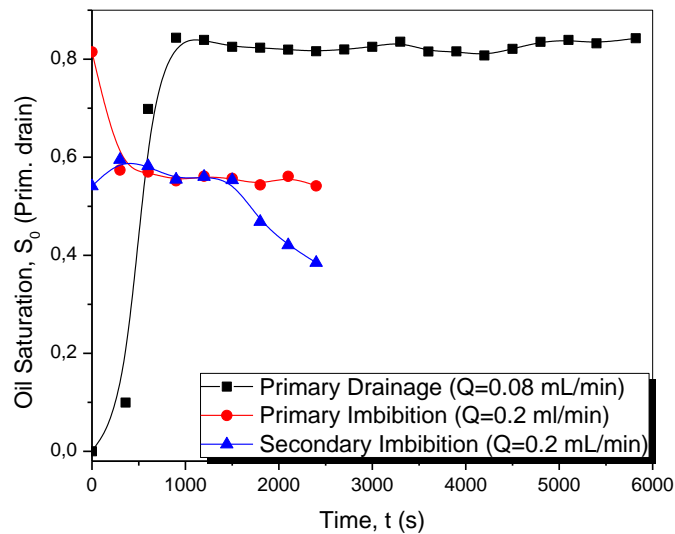
relatively low EOR efficiency was achieved with the PNP dispersion

Successive snap-shots of the displacement of residual paraffin oil (secondary imbibition) by:

(a)

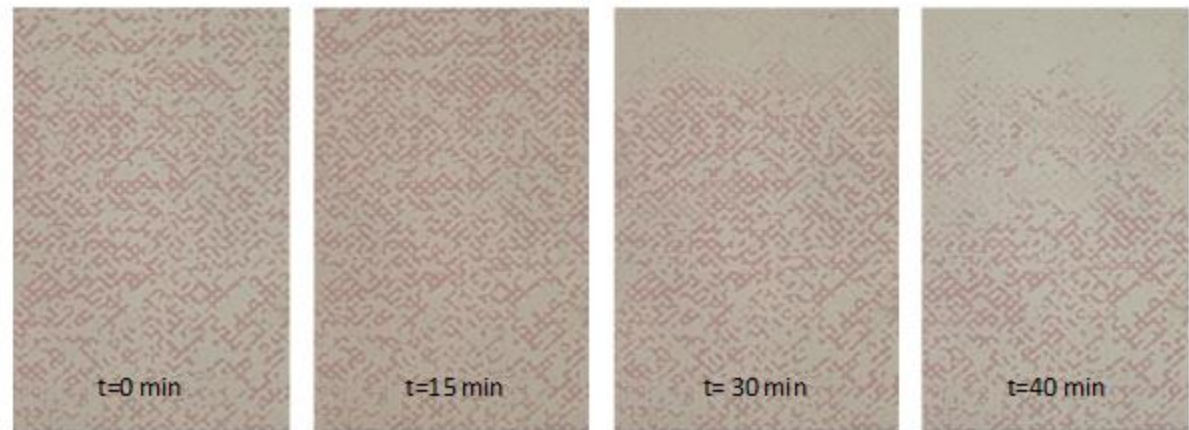


**(b) Pickering emulsion along with PNP dispersion**



significant EOR efficiency achieved by emulsion and PNP dispersion at equal flow rates

(b)



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## Displacement tests conducted on the pore network model

Type of displacement	Displaced fluid	Displacing fluid	Flow rate (mL/min)	Injected volume (mL)	Oil saturation	Oil removal efficiency (%)
Drainage	Brine*	Paraffin oil	0.08	8.0	0.825	-
Prim. Imbib.	Resid. paraffin oil	Brine*	0.2	8.0	0.557	32.5
Sec. Imbib.	Resid. paraffin oil	PNP dispersion**	0.2	8.0	0.546	1.97
Drainage	Brine*	Paraffin oil	0.08	8.0	0.843	-
Prim. Imbib.	Resid. paraffin oil	Brine*	0.2	8.0	0.541	35.8
Sec. Imbib.	Resid. paraffin oil	Emulsion***	0.1	4.02	0.385	28.8
		PNP dispersion*	0.1	4.0		

\*0.5M NaCl / 0.25M CaCl<sub>2</sub>; \*\*0.25% SiO<sub>2</sub>-P(AMPSA-co-DMA) / 0.5M NaCl / 0.25M CaCl<sub>2</sub>; \*\*\*PNP dispersion / n-C<sub>10</sub>



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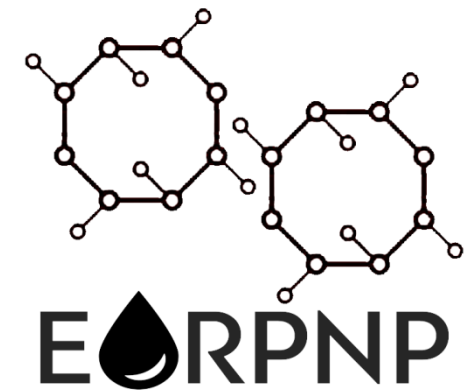
## Conclusions

- Successful **surface functionalization of SiO<sub>2</sub> NPs by MPS and P(AMPSA-co-DMA) copolymer**
- **Surface tension** depends on the concentration of SiO<sub>2</sub>-P(AMPSA-co-DMA) PNPs
  - with the addition of NaCl, the ST changes weakly
  - with the addition of CaCl<sub>2</sub>, the ST drops significantly
- **Relatively low EOR efficiency** was achieved with the SiO<sub>2</sub>-P(AMPSA-co-DMA)0.25% /NaCl 0.5M /CaCl<sub>2</sub> 0.25M PNP dispersion
- **Significant EOR efficiency** achieved by injecting simultaneously Pickering emulsion and PNP dispersion
- Better displacement of paraffin oil was exhibited by the Pickering emulsion along with the PNP dispersion: the presence of the viscous emulsion facilitates the displacement of the viscous paraffin oil

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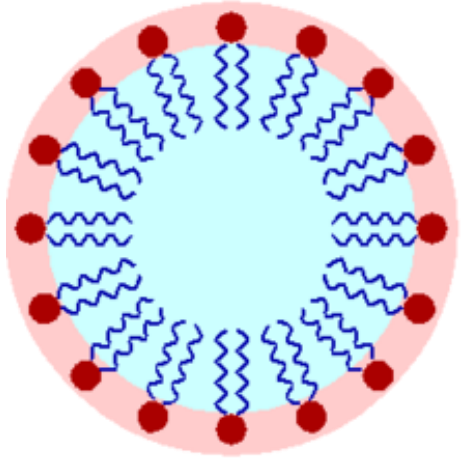
## Acknowledgments

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Thank you for  
your attention!!!

